

# CORE Chemistry topics

## 2025-2026 School Year

| Unit                                | Days | Dates          | Workbook pages | Standards & Key Ideas (NYS Chemistry Core Curriculum)  |
|-------------------------------------|------|----------------|----------------|--|
| <b>Atomic Structure</b>             | 6    | Sept 3–10      | 1-24           | <i>Key Idea 2.1:</i> Matter is made up of particles; atoms are the smallest unit. <i>Key Idea 3.1:</i> Modern atomic theory; nucleus, electrons, orbitals. |
| <b>Periodic Table</b>               | 6    | Sept 11-19     | 77-96          | <i>Key Idea 3.1f–3.1j:</i> Periodic trends (electronegativity, ionization energy, atomic radius), classification of elements.                              |
| <b>Bonding</b>                      | 6    | Sept 22-Oct 1  | 97-116         | <i>Key Idea 5.2:</i> Chemical bonding (ionic, covalent, metallic, intermolecular forces), physical properties related to bonding.                          |
| <b>Chem-Math</b>                    | 9    | Oct 3-16       | 45-56          | <i>Key Idea 1.1c, 1.2, 3.2:</i> Mole calculations, percent composition, empirical/molecular formulas, dimensional analysis.                                |
| <b>Matter &amp; Energy</b>          | 9    | Oct 17-29      | 57-76          | <i>Key Idea 2.1, 2.2, 2.3, 3.2:</i> Phases of matter, physical/chemical changes, energy transfer, heat, specific heat, gas laws.                           |
| <b>Solutions</b>                    | 5    | Oct 30 - Nov 6 | 117-134        | <i>Key Idea 2.2, 3.3:</i> Concentration (molarity, ppm), solubility, factors affecting rate of dissolving, colligative properties.                         |
| <b>Kinetics &amp; Equilibrium</b>   | 7    | Nov 7-21       | 135-156        | <i>Key Idea 3.4, 5.2:</i> Collision theory, reaction rates, reversible reactions, Le Chatelier's Principle, entropy, enthalpy.                             |
| <b>Acids &amp; Bases</b>            | 6    | Nov 24-Dec3    | 173-192        | <i>Key Idea 3.5:</i> pH, properties of acids/bases, indicators, neutralization, titrations, Arrhenius vs. Bronsted definitions.                            |
| <b>Redox &amp; Electrochemistry</b> | 6    | Dec 4-10       | 157-172        | <i>Key Idea 3.4, 3.6:</i> Oxidation numbers, redox reactions, electrochemical and electrolytic cells, half-reactions, voltaic cells.                       |
| <b>Organic Chemistry</b>            | 7    | Dec 11-19      | 193-214        | <i>Key Idea 4.1:</i> Organic functional groups, naming, reactions (substitution, addition, esterification, polymerization).                                |

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|--------------------------|---|-----------|---------|---|
| <b>Nuclear Chemistry</b> | 5 | Jan 5-9   | 215-232 | <i>Key Idea 3.7:</i> Types of radiation, nuclear decay, fission/fusion, half-life calculations, risks and benefits of nuclear energy. |
| <b>Regents Review</b>    | 5 | Jan 12-16 |         |   |
| <b>Regents Week</b>      |   | Jan 20-23 |         |   |