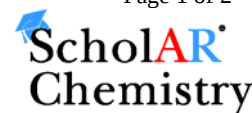


Iron (III) Nitrate Solution, 2.0M



MSDS # 376.50

Section 1: Product and Company Identification

Iron (III) Nitrate Solution, 2.0M

Synonyms/General Names: Ferric Nitrate, Nonahydrate**Product Use:** For educational use only**Manufacturer:** Columbus Chemical Industries, Inc., Columbus, WI 53925.

24 Hour Emergency Information Telephone Numbers

CHEMTREC (USA): 800-424-9300**CANUTEC (Canada): 613-424-6666**

Scholar Chemistry; 5100 W. Henrietta Rd, Rochester, NY 14586; (866) 260-0501; www.Scholarchemistry.com

Section 2: Hazards Identification

Clear, yellow liquid; no odor.

WARNING! Oxidizing agent and severe body tissue irritant

Target organs: Blood

This material is considered hazardous by the OSHA Hazard Communication Standard (29 CFR 1910.1200).

HMIS (0 to 4)

Health	1
Fire Hazard	0
Reactivity	2

Section 3: Composition / Information on Ingredients

Ferric Nitrate, Nonahydrate (7782-61-8), 34-38%.

Water (7732-18-5), 62-66%.

Section 4: First Aid Measures

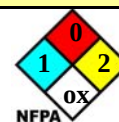
*Always seek professional medical attention after first aid measures are provided.***Eyes:** Immediately flush eyes with excess water for 15 minutes, lifting lower and upper eyelids occasionally.**Skin:** Immediately flush skin with excess water for 15 minutes while removing contaminated clothing.**Ingestion:** Call Poison Control immediately. Rinse mouth with cold water. Give victim 1-2 cups of water or milk to drink. Induce vomiting immediately.**Inhalation:** Remove to fresh air. If not breathing, give artificial respiration.

Section 5: Fire Fighting Measures

Oxidizing agent. When heated to decomposition, emits acid fumes.

Protective equipment and precautions for firefighters: Use foam or dry chemical to extinguish fire.

Firefighters should wear full fire fighting turn-out gear and respiratory protection (SCBA). Cool container with water spray. Material is not sensitive to mechanical impact or static discharge.



Section 6: Accidental Release Measures

Use personal protection recommended in Section 8. Isolate the hazard area and deny entry to unnecessary and unprotected personnel. Contain spill with sand or absorbent material and place in sealed bag or container for disposal. Ventilate and wash spill area after pickup is complete. See Section 13 for disposal information.
13 for disposal information.

Section 7: Handling and Storage

Yellow**Handling:** Use with adequate ventilation and do not breathe dust or vapor. Avoid contact with skin, eyes, or clothing. Wash hands thoroughly after handling.**Storage:** Store in Oxidizer Storage Area [Yellow Storage] with other oxidizers and away from any combustible materials. Store in a cool, dry, well-ventilated, locked store room away from incompatible materials.

Section 8: Exposure Controls / Personal Protection

Use ventilation to keep airborne concentrations below exposure limits. Have approved eyewash facility, safety shower, and fire extinguishers readily available. Wear chemical splash goggles and chemical resistant clothing such as gloves and aprons. Wash hands thoroughly after handling material and before eating or drinking. Exposure guidelines: Iron Salts, soluble: OSHA PEL: Not Available, ACGIH: TLV: 1 mg/m³, STEL: Not Available.

Section 9: Physical and Chemical Properties

Molecular formula	Fe(NO ₃) ₃ •9H ₂ O.	Appearance	Clear, yellow liquid.
Molecular weight	404.00.	Odor	No odor.
Specific Gravity	1.30 g/mL @ 20°C.	Odor Threshold	N/A.
Vapor Density (air=1)	N/A.	Solubility	Soluble in water and alcohol .
Melting Point	N/A.	Evaporation rate	N/A. (Butyl acetate = 1).
Boiling Point/Range	N/A.	Partition Coefficient	N/A. (log P _{OW}).
Vapor Pressure (20°C)	N/A.	pH	N/A.
Flash Point:	N/A.	LEL	N/A.
Autoignition Temp.:	N/A.	UEL	N/A.

N/A = Not available or applicable

Section 10: Stability and Reactivity

Stability: Stable under normal conditions of use and storage.

Incompatibility: Aluminums, cyanides, phosphorous, stannous chloride, thiocyanates, oxidizing materials, organic materials and sodium hypophosphite.

Shelf life: Indefinite shelf life, store in cool, dry environment.

Section 11: Toxicology Information

Acute Symptoms/Signs of exposure: **Eyes:** Redness, tearing, itching, burning, conjunctivitis. **Skin:** Redness, itching.

Ingestion: Irritation and burning sensations of mouth and throat, nausea, vomiting and abdominal pain. **Inhalation:** Irritation of mucous membranes, coughing, wheezing, shortness of breath,

Chronic Effects: No information found.

Sensitization: none expected

Iron (III) Nitrate 9-hydrate: LD50 [oral, rat]; 3250mg/kg; LC50 [rat]; N/A; LD50 Dermal [rabbit]; N/A

Material has not been found to be a carcinogen nor produce genetic, reproductive, or developmental effects.

Section 12: Ecological Information

Ecotoxicity (aquatic and terrestrial): Ecological impact has not been determined.

Section 13: Disposal Considerations

Check with all applicable local, regional, and national laws and regulations. Local regulations may be more stringent than regional or national regulations. Small amounts of this material may be suitable for sanitary sewer or trash disposal.

Section 14: Transport Information

DOT Shipping Name:	Ferric Nitrate solution.	Canada TDG:	Ferric Nitrate solution.
DOT Hazard Class:	5.1, pg III.	Hazard Class:	5.1, pg III.
Identification Number:	UN1466.	UN Number:	UN1466.

Section 15: Regulatory Information

EINECS: Not listed.

TSCA: All components are listed or are exempt.

WHMIS Canada: CLASS C: Oxidizing material .

California Proposition 65: Not listed.

The product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all the information required by the Controlled Products Regulations.

Section 16: Other Information

Current Issue Date: January 23, 2009

Disclaimer: Scholar Chemistry and Columbus Chemical Industries, Inc., ("S&C") believes that the information herein is factual but is not intended to be all inclusive. The information relates only to the specific material designated and does not relate to its use in combination with other materials or its use as to any particular process. Because safety standards and regulations are subject to change and because S&C has no continuing control over the material, those handling, storing or using the material should satisfy themselves that they have current information regarding the particular way the material is handled, stored or used and that the same is done in accordance with federal, state and local law. S&C makes no warranty, expressed or implied, including (without limitation) warranties with respect to the completeness or continuing accuracy of the information contained herein or with respect to fitness for any particular use.