

Week-At-A-Glance

Date: 9-18 to 9-22

	Monday	Tuesday	Wednesday	Thursday	Friday
Standard/Objective	8.P.1.3 Compare physical changes such as size, shape, and state to chemical changes that are the result of a chemical reaction to include changes in color, temperature, formation of a gas or precipitate.	8.P.1.3 Compare physical changes such as size, shape, and state to chemical changes that are the result of a chemical reaction to include changes in color, temperature, formation of a gas or precipitate.	½ Day 8.P.1.3 Compare physical changes such as size, shape, and state to chemical changes that are the result of a chemical reaction to include changes in color, temperature, formation of a gas or precipitate.	8.P.1.2 Explain how the physical properties of elements and their reactivity have been used to produce the current model of the Periodic Table of elements.	8.P.1.2 Explain how the physical properties of elements and their reactivity have been used to produce the current model of the Periodic Table of elements.
Learning Target	I can write notes on pH in order to understand how pH is used to classify solutions.	I can investigate how carbon dioxide changes the pH of a solution.	I can review concepts related to matter.	I can acquire knowledge about subatomic particles.	I can apply my understanding of subatomic particles to specific elements and create models

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					of those elements.
Assignments/Activities	Bell Ringer Cornell Notes on pH	Bell Ringer Vernier CO2 Lab	Quizlet Live	Bell Ringer Atoms Family Notes	Bell Ringer Bohr Model of an Atom activity Phet Lab Simulator (creating elements)
Graded Assessments and/or projects		Vernier CO2 Lab			Bohr Model for 2 elements
Homework	Complete assignments if absent	Complete assignments if absent	Complete assignments if absent	Complete assignments if absent	Complete assignments if absent