## Week-At-A-Glance

## Date: \_\_\_\_9-18 to 9-22\_\_\_\_\_

	Monday	Tuesday	Wednesday	Thursday	Friday
	8.P.1.3	8.P.1.3	½ Day	8.P.1.2 Explain	8.P.1.2 Explain
Standard/Objective	Compare	Compare		how the	how the
	physical	physical	8.P.1.3	physical	physical
	changes such	changes such	Compare	properties of	properties of
	as size, shape,	as size, shape,	physical	elements and	elements and
	and state to	and state to	changes such	their reactivity	their reactivity
	chemical	chemical	as size, shape,	have been	have been
	changes that	changes that	and state to	used to	used to
	are the result	are the result of	chemical	produce the	produce the
	of a chemical	a chemical	changes that	current model	current model
	reaction to	reaction to	are the result of	of the Periodic	of the Periodic
	include	include	a chemical	Table of	Table of
	changes in	changes in	reaction to	elements.	elements.
	color,	color,	include		
	temperature,	temperature,	changes in		
	formation of a	formation of a	color,		
	gas or	gas or	temperature,		
	precipitate.	precipitate.	formation of a		
			gas or		
			precipitate.		
<u>-</u> .	I can write	lcan	l can review	l can acquire	I can apply my
Learning Target	notes on pH in	investigate how	concepts	knowledge	understanding
	order to	carbon dioxide	related to	about .	of subatomic
	understand	changes the	matter.	subatomic	particles to
	how pH is	pH of a		particles.	specific
	used to	solution.			elements and
	classify				create models
	solutions.				

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					of those elements.
Assignments/Activities	Bell Ringer	Bell Ringer	Quizlet Live	Bell Ringer	Bell Ringer
	Cornell Notes on pH	Vernier CO2 Lab		Atoms Family Notes	Bohr Model of an Atom activity
					Phet Lab Simulator (creating elements)
Graded Assessments and/or projects		Vernier CO2 Lab			Bohr Model for 2 elements
Homework	Complete assignments if absent				