Week-At-A-Glance

Date: ___9-25 to 9-29_____

	Monday	Tuesday	Wednesday	Thursday	Friday
Standard/Objective	8.P.1.3 Compare physical changes such as size, shape, and state to chemical changes that are the result of a chemical reaction to include changes in color, temperature, formation of a gas or precipitate.	8.P.1.4 Explain how the idea of atoms and a balanced chemical equation support the law of conservation of mass.	8.P.1.4 Explain how the idea of atoms and a balanced chemical equation support the law of conservation of mass.	8.P.1.4 Explain how the idea of atoms and a balanced chemical equation support the law of conservation of mass.	8.P.1.4 Explain how the idea of atoms and a balanced chemical equation support the law of conservation of mass.
Learning Target	I can apply my understanding of the different types of mixtures to Ms. Yencha's demonstration in order to explain	I can understand the law of conservation and how balanced chemical equations are an example.	I can practice balancing chemical equations.	I can practice balancing chemical equations.	I can practice balancing chemical equations.

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	physical properties.				
Assignments/Activities	Bell Ringer Homogenous versus heterogenous mixtures activity Notes on physical properties	Bell Ringer Law of conservation and balancing equations cornell notes Exit Ticket	Bell Ringer Balancing Equations Practice Foldable	Bell Ringer Balancing Equations Practice Foldable	Bell Ringer Quiz Balancing Equations Practice Foldable
Graded Assessments and/or projects		Exit Ticket		AVID notebook check. Students should have notes on the states of matter, the categorization of matter, physical vs chemical changes, pH, The Atoms Family, Physical	Quiz

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				Properties, Law of conservation, and Balancing chemical equations	
Homework	Complete assignments if absent	Complete assignments if absent	Complete assignments if absent	Complete assignments if absent	Complete assignments if absent