

| State of Michigan Recognized Grade Level Content Expectations (GLCEs) |  |   |   |   |   |
|---|--|---|---|---|---|
| Timeline  |  |   |   |   |   |
| <b>Week 1-3</b>   | <b>Unit 1: Scientific Method/Process</b>   |   |   |   |   |
|   | <b>Objective: Understand, Use, and Apply the Scientific Method/Process</b>   |   | <b>Text Resources</b><br>(by book ID on book spine) | <b>Vocabulary</b>   | <b>Assessment</b>                                     |
|   | <b>S.IP.07.11</b>  | I can write scientific questions based on my observations, investigations, and research.  | K: 128, A 8-13                                      | hypothesis, variable, independent, dependent, control, chart, graph, data, scientific method, metric system, mass, volume, density, temperature, Celsius, gram, liter, meter, weight, trends/pattern in data  | Design experiment based on Scientific Method          |
|   | <b>S.IP.07.12</b>  | I can create and run scientific investigations.   | K: 131-136  |   |   |
|   | <b>S.IP.07.13</b>  | I can use tools and equipment (spring scales, stop watches, meter sticks and tapes, models, hand lenses, thermometers, models, sieves, microscopes, hot plates, pH meters) needed for scientific investigations.              | K: 133-135  |   |   |
|   | <b>S.IP.07.15</b>  | I can make charts and graphs from data and observations.  | K: 131, 159-160                                     |   |   |
|   | <b>S.IP.07.16</b>  | I can identify patterns in data   | K: 135-136  |   |   |
|   | <b>S.IA.07.11</b>  | I can make sense of information from data tables and graphs to answer scientific questions.   | K: 159-160  |   |   |
|   | <b>S.IA.07.14</b>  | I can use data from several experiments when determining a valid conclusion.  | Additional resources in Schoology                   |   |   |
|   | <b>S.RS.07.11</b>  | I can evaluate the strengths & weaknesses of data   |   |   |   |
|   | <b>S.RS.07.14</b>  | I can evaluate scientific explanations  |   |   |   |
|   | <b>S.RS.07.15</b>  | I can show scientific concepts through activities   |   |   |   |
|   |  | I can write a Claim, Evidence and provide Reasoning   |   |   |   |
| <b>7 Weeks</b>  | <b>Unit 2: Waves &amp; Energy</b>  |   |   |   |   |
|   | <b>Objective A: Using sound waves, seismic waves, waves in water, and light waves, demonstrate how waves transfer energy.</b>  |   |   |   |   |
|   | <b>Objective B: Describe how the sun is the major source of light and heat on Earth.</b>   |   |   |   |   |
|   | <b>Objective C: Demonstrate how only a tiny fraction of the light energy from the sun reaches Earth to heat it.</b>  |   | <b>Text Resources</b><br>(by book ID on book spine) | <b>Vocabulary</b>   | <b>Assessment</b>                                     |
|   | <b>P.EN.07.31</b>  | I can give examples of waves, including sound, seismic, and water waves.  | O: 6-37   | waves, wavelength, transverse waves, compressional waves, seismic waves, light wave, water wave, sound wave, crests, troughs, amplitude, frequency, wavelength, light energy, sound wave, energy transfer, energy, vibration,   | Unit 2 Pre-Post Test<br>Vocab Quizzes<br>Content Quiz |
|   | <b>P.EN.07.32</b>  | I can show how waves are produced by vibrations in matter.  | O: 11, 36-37  |   |   |
|   | <b>P.EN.07.33</b>  | I can show by example how waves transfer energy when they interact with matter (for example tuning fork in water, waves hitting a beach, earthquake knocking over buildings).   | O: 8-9, 66  |   |   |
|   | <b>P.EN.07.61</b>  | I can recognize what a nuclear reaction is and that it takes place in the sun, producing heat and light.  |   |   |   |
|   | <b>P.EN.07.62</b>  | I can talk about how only a tiny fraction of light energy from the sun becomes heat energy on Earth.<br>* All Resources located in Schoology  | O: 74-75<br>Additional resources in Schoology       |   |   |
| <b>8 Weeks</b>  | <b>Unit 3: Physical and Chemical Properties and Changes in Matter</b>  |   |   |   |   |
|   | <b>Unit 3A: Elements and Compounds</b>   |   |   |   |   |
|   | <b>Objective A: Identify elements as the chemical substances that make up all other substances and are composed of one kind of atom.</b>   |   |   |   |   |
|   | <b>Objective B: Elements are organized on the Periodic Table in families.</b>  |   |   |   |   |
|   | <b>Objective C: Describe examples of physical and chemical properties of elements and compounds.</b>   |   | <b>Text Resources</b><br>(by book ID on book spine) | <b>Vocabulary</b>   | <b>Assessment</b>                                     |
|   | <b>P.PM.07.21</b>  | I can explain how an atom as the smallest component making up an element.   | K: 8-14   | atom, atomic arrangement, chemical change, chemical properties (of elements or compounds), chemical reaction, molecule, nonmetal reactive gases, products, reactants, density, boiling point, conductivity, pH scale (measured with paper/meter), elements, periodic table of the elements, physical change, compound, classification of substances, conservation of mass, graduated cylinder, physical properties (of elements or compounds), phase change, pH | Unit 3 Pre-Post Test<br>Vocab Quizzes<br>Content Quiz |
|   | <b>P.PM.07.23</b>  | I can draw or make a model showing the structure of molecules (water, carbon dioxide, salt).  | L: 16-24, A 70                                      |   |   |
|   | <b>P.PM.07.22</b>  | I can tell how the elements on the Periodic Table are organized by similar properties into families (highly reactive metals, less reactive metals, highly reactive nonmetals, and some almost completely non-reactive gases). | K: 98-116   |   |   |
|   | <b>P.PM.07.11</b>  | I can group substances by their chemical properties (flammability, pH, acid-base indicators, reactivity).   | K: 76   |   |   |
|   | <b>P.PM.07.24</b>  | I can list traits of elements and compounds and tell whether the traits are physical (such as boiling point, density, or color) or chemical (such as conductivity and reactivity).  | K: 72-76  |   |   |
|   |  | * All Resources located in Schoology  | Additional resources in Schoology                   |   |   |
|   | <b>Unit 3B: Chemical Properties</b>  |   |   |   |   |
|   | <b>Objective A: Identify evidence of a chemical change through color, gas formation, solid formation, and temperature change</b>   |   |   |   |   |
|   | <b>Objective B: Compare and contrast the properties of reactants and products in a chemical change, using changes in properties as evidence of the change. (implied: define chemical properties)</b> |   |   |   |   |
|   | <b>Objective C: Illustrate the structure of molecules using models or drawings.</b>  |   | <b>Text Resources</b><br>(by book ID on book spine) | <b>Vocabulary</b>   | <b>Assessment</b>                                     |
|   | <b>Objective D: Demonstrate that in a chemical change mass is conserved.</b>   |   |   |   |   |
|   | <b>P.CM.07.21</b>  | I can recognize a chemical change from change in color, formation of a gas, formation of a solid, and change in temperature.  | K: 78-87  | reactants, products, conservation of matter/mass, products, reactants,  | Unit 3 Pre-Post Test                                  |



