

Electric Charge

1. A lightning bolt has a charge of 20 C. How many electrons make up the bolt?

$$20 = n(1.602 \times 10^{-19}) \quad n = 1.2 \times 10^{20}$$

2. An electroscope has a charge of 1.2 μC . How many electrons pass through your fingers as you ground this electroscope?

$$1.2 \times 10^{-6} = n(1.602 \times 10^{-19}) \quad n = 7.5 \times 10^{12}$$

3. A charged rod has an excess of 6.4×10^8 electrons which it shares equally with a pith ball when they touch. What is the charge on the pith ball?

$$Q = 3.2 \times 10^8 (1.602 \times 10^{-19}) = 51 \times 10^{-11} \text{ C}$$

4. Calculate the electrostatic force between a proton and an electron which are 1×10^{-10} m apart.

$$F = \frac{(9 \times 10^9)(1.602 \times 10^{-19})(-1.602 \times 10^{-19})}{(1 \times 10^{-10})^2} = 2.3 \times 10^{-8} \text{ N}$$

5. A pith ball with a charge of $+6.0 \mu\text{C}$ is placed 12 cm from another ball with a charge of $-4.3 \mu\text{C}$.

- a) Which ball has an excess of electrons?

$$-4.3 \mu\text{C}$$

- i) How many electrons in excess does it have?

$$4.3 \times 10^{-6} = n(1.602 \times 10^{-19}) \quad n = 2.7 \times 10^{13}$$

- b) Which ball has a deficit of electrons?

$$6.0 \mu\text{C}$$

- ii) How many electrons in deficit does it have?

$$6.0 \times 10^{-6} = n(1.602 \times 10^{-19}) \quad n = 3.7 \times 10^{13}$$

- c) What is the force acting between the two balls?

$$F = \frac{(9 \times 10^9)(4.3 \times 10^{-6})(6.0 \times 10^{-6})}{(0.12)^2} = 16 \text{ N}$$

- d) Is the force attractive or repulsive?

6. A charge of $17.0 \mu\text{C}$ is placed 15.0 cm from a second charge. The force of attraction between the two is 21.4 N. Calculate the second charge.

$$21.4 = \frac{(9 \times 10^9)(17 \times 10^{-6})Q_2}{(0.15)^2} \quad Q_2 = 3.15 \times 10^{-6} \text{ C}$$

7. a) Calculate the electric force holding an electron in orbit ($r = 0.53 \times 10^{-10}$ m) around a proton.

$$F = \frac{(9.0 \times 10^9)(1.602 \times 10^{-19})(1.602 \times 10^{-19})}{(0.53 \times 10^{-10})^2} = 8.2 \times 10^{-8} \text{ N}$$

- b) Calculate the gravitational force between the same proton and electron.
(mass of electron = 9.11×10^{-31} kg and mass of proton = 1.67×10^{-27} kg)

$$F_G = \frac{(6.67 \times 10^{-11})(9.11 \times 10^{-31})(1.67 \times 10^{-27})}{(0.53 \times 10^{-10})^2} = 3.6 \times 10^{-47} \text{ N}$$

c) Which force is more significant within the atom, electrostatic or gravitational?

8. How many electrons make up a charge of $30.0 \mu\text{C}$?

$$30 \times 10^{-6} = n (1.602 \times 10^{-19}) \quad n = 1.9 \times 10^{14}$$

9. Two charged smoke particles exert a force of 0.042 N on each other. What will the force if they are moved so they are only half as far apart?

~~0.042 N~~

4x greater

0.168 N

10. What is the magnitude of the electric force of attraction between an iron nucleus (26 protons) and its innermost electron if the distance between them is $1.5 \times 10^{-12} \text{ m}$?

$$F = \frac{(9 \times 10^9) (26) (1.602 \times 10^{-19})^2}{(1.5 \times 10^{-12})^2} = 0.0027 \text{ N}$$

11. A person scuffing her feet on a wool rug on a dry day accumulates a net charge of $-60 \mu\text{C}$. How many excess electrons does this person get, and by how much does her mass increase?

$$60 \times 10^{-6} = n (1.602 \times 10^{-19})$$

$$3.75 \times 10^{14} e \times \frac{9.11 \times 10^{-31} \text{ kg}}{1 e} = 3.4 \times 10^{-16} \text{ kg}$$